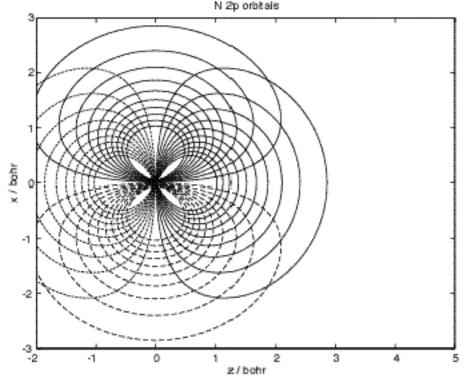


And of the  $\pi$  non-bonding orbitals (the  $2p_x$  and  $2p_y$  on N) N 2p orbitals

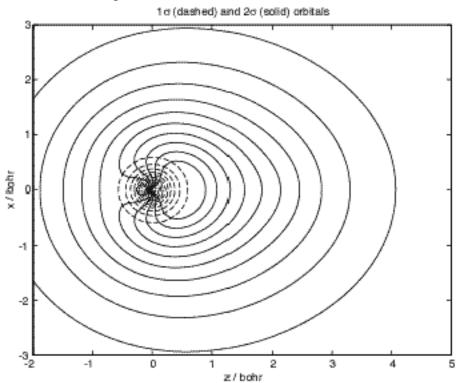


МО	$1s_{N}$	$2s_{\rm N}$	2pz	$1s_{\rm H}$
1σ	0.99133	0.045212	0.007862	-0.01686
$2\sigma$	-0.21017	0.83545	0.16467	0.23935
3σ	0.10336	-0.58975	0.66431	0.49299
$4\sigma$	0.089038	-0.75158	-0.94876	1.2744

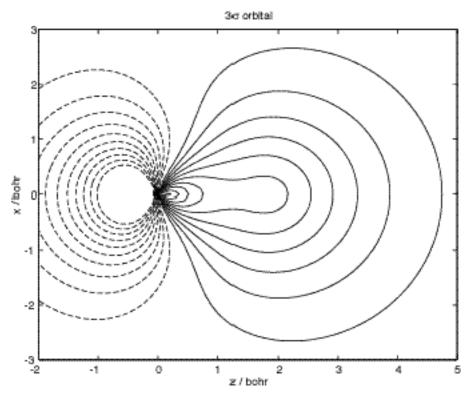
The NH  $\sigma$  molecular orbitals in the  ${}^{3}\Sigma^{-}$  state are linear combinations of these atomic orbitals

Note, that the H 1s orbital makes a contribution to both the  $2\sigma$  and  $3\sigma$  bonding orbitals. Nevertheless, overall one can say the the  $2\sigma$  orbital is primarily the N 2s orbital while the  $3\sigma$  orbital is a linear combination of the 2s and 2p orbitals on N and the 1s orbital on H. The single NH  $\pi$  orbital is identical to the N 2p orbital.

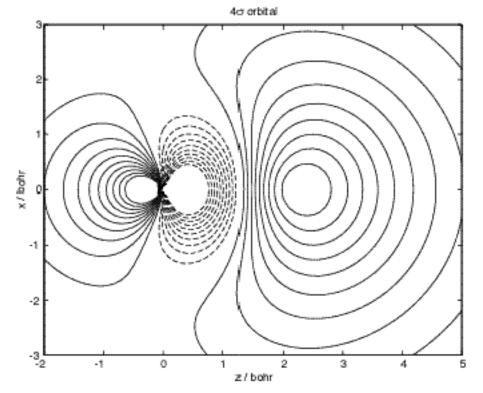
Here is a contour plot of the  $1\sigma$  and  $2\sigma$  orbitals

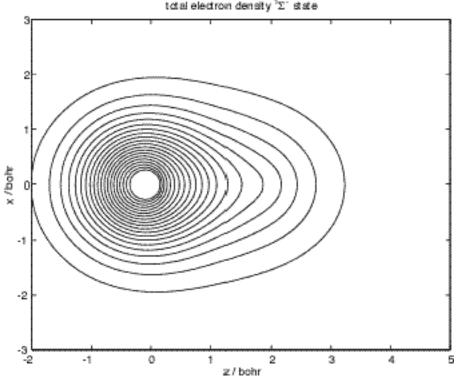


Here is the  $3\sigma$  orbital



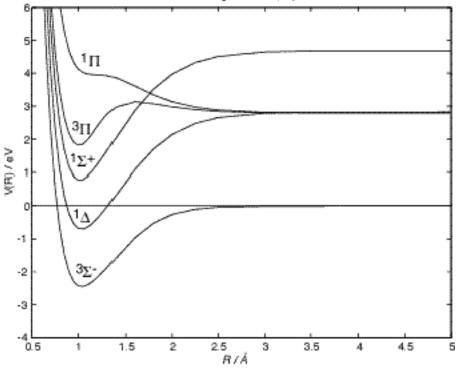
and the anti-bonding (you can see the node between the atoms)  $4\sigma$  orbital



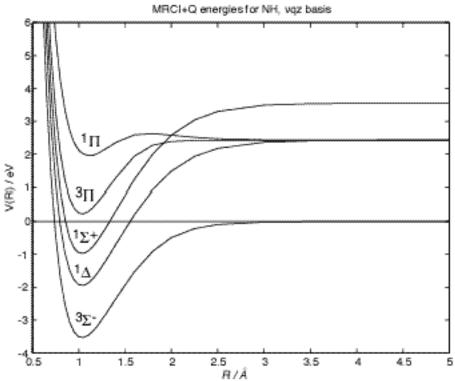


Here is the total electron density (the one-particle density matrix) for NH in the  ${}^{3}\Sigma^{-}$  state total electron density  ${}^{3}\Sigma^{-}$  state

Here, now, is a plot of the multi-configuration SCF energies for the lowest states of NH MCSCF energies for NH, vqz basis



The  ${}^{3}\Sigma^{-}$ ,  ${}^{1}\Delta$ , and  ${}^{1}\Sigma^{+}$  states all correspond to the  $1\sigma^{2}2\sigma^{2}3\sigma^{2}1\pi^{2}$  electronic configuration, while the  ${}^{1,3}\Pi$  states correspond to  $1\sigma^{2}2\sigma^{2}3\sigma^{1}1\pi^{3}$ . When configuration interaction is included, to account for dynamic electron correlation, the potential curves are all lowered, as shown in the next figure. Note that the curves go, asymptotically, to the expected atomic limits [ N({}^{4}S, {}^{2}D, {}^{2}P)+H].



The calculated spectroscopic properties of the individual states are shown in the next table. This is pretty good agreement!

State	$D_o(eV)$		$T_e (\mathrm{cm}^{-1})$		<i>r</i> <sub>e</sub> (Å)	
	MRCI+Q	exp	MRCI+Q	exp	MRCI+Q	exp
3 <sub>2</sub> -	3.33	3.40			1.0318	1.0362
$^{1}\Delta$			12741	(12566)	1.0353	1.034
$1\Sigma^+$			20587	21202	1.0342	1.036
<sup>3</sup> П			30290	29807	1.0359	1.0369
1П			44811	(43744)	1.0079	1.110

Values in parentheses are known with less certainty.